## Pearson Edexcel

# Mark Scheme (Provisional) 

## Summer 2021

Pearson Edexcel International GCSE
In Physics (4PH1) Paper 1P and Science (Double Award) (4SD0) Paper 1P

## Edexcel and BTEC Qualifications

Edexcel and BTEC qualifications are awarded by Pearson, the UK's largest awarding body. We provide a wide range of qualifications including academic, vocational, occupational and specific programmes for employers. For further information visit our qualifications websites at www.edexcel.com or www.btec.co.uk. Alternatively, you can get in touch with us using the details on our contact us page at www.edexcel.com/contactus.

## Pearson: helping people progress, everywhere

Pearson aspires to be the world's leading learning company. Our aim is to help everyone progress in their lives through education. We believe in every kind of learning, for all kinds of people, wherever they are in the world. We've been involved in education for over 150 years, and by working across 70 countries, in 100 languages, we have built an international reputation for our commitment to high standards and raising achievement through innovation in education. Find out more about how we can help you and your students at: www.pearson.com/uk

Summer 2021
Question Paper Log Number 67159
Publications Code 4SD0_1P_2106_MS
All the material in this publication is copyright
© Pearson Education Ltd 2021

## General Marking Guidance

- All candidates must receive the same treatment. Examiners must mark the first candidate in exactly the same way as they mark the last.
- Mark schemes should be applied positively. Candidates must be rewarded for what they have shown they can do rather than penalised for omissions.
- Examiners should mark according to the mark scheme not according to their perception of where the grade boundaries may lie.
- There is no ceiling on achievement. All marks on the mark scheme should be used appropriately.
- All the marks on the mark scheme are designed to be awarded. Examiners should always award full marks if deserved, i.e. if the answer matches the mark scheme. Examiners should also be prepared to award zero marks if the candidate's response is not worthy of credit according to the mark scheme.
- Where some judgement is required, mark schemes will provide the principles by which marks will be awarded and exemplification may be limited.
- When examiners are in doubt regarding the application of the mark scheme to a candidate's response, the team leader must be consulted.
- Crossed out work should be marked UNLESS the candidate has replaced it with an alternative response.

| Question <br> number | Answer | Notes | Marks |
| :---: | :--- | :--- | :---: |
| 1 (a) | C (white); <br> A is incorrect because its temperature is the second <br> lowest <br> B is incorrect because its temperature is the lowest <br> D is incorrect because its temperature is the second <br> highest |  | 1 |
| (ii) | D (white dwarf); <br> A is incorrect because the Sun is not massive <br> enough to form a black hole <br> B is incorrect because the Sun is not massive <br> enough to form a neutron star <br> C is incorrect because the Sun is not massive <br> enough to form a supernova <br> C (supernova); <br> A is incorrect because this is during the middle of <br> the life cycle <br> B is incorrect because this is an early stage of the <br> life cycle <br> D is incorrect because this is a late stage in the life <br> cycle of stars like our Sun |  |  |


| Question number | Answer | Notes | Marks |
| :---: | :---: | :---: | :---: |
| 2 (a) | (plotting) compass(es); | allow suspended magnet, magnetometer allow higher level responses e.g. Hall probe, search coil | 1 |
| (b) | one mark for each correct indication;;; | 2 marks max. if more than three indications given | 3 |
| (c) | any two from: difficult to magnetise; difficult to demagnetise; <br> idea that it retains its magnetism; | allow idea of taking a long time to magnetise allow idea of taking a long time to demagnetise | 2 |

Total for Question 2 = 6 marks

| Question number | Answer | Notes | Marks |
| :---: | :---: | :---: | :---: |
| 3 (a) (i) <br> (ii) <br> (iii) | neutron; <br> nucleus/nuclei splitting; <br> releasing (two) daughter nuclei / neutrons / energy; <br> neutrons released (by fission) are absorbed by other (uranium) nuclei; causing fission/splitting in other (uranium) nuclei; | condone nucleus/nuclei breaking apart ignore daughter cells allow smaller/lighter nuclei for daughter nuclei <br> condone atoms for nuclei condone atoms for nuclei | 1 <br> 2 <br> 2 |
| (b) | concrete / lead / (thick) steel; |  | 1 |
| (c) | graphite; slow; boron; absorb; |  | 4 |

Total for Question 3 = 10 marks

| Question number | Answer | Notes | Marks |
| :---: | :---: | :---: | :---: |
| 4 (a) (i) | use of acceleration = change in velocity / time; <br> substitution; <br> evaluation; <br> e.g. <br> acceleration $=$ change in velocity $/$ time <br> acceleration $=(-) 30 / 6.2$ <br> (acceleration $=)(-) 4.8\left(\mathrm{~m} / \mathrm{s}^{2}\right)$ | seen anywhere in working allow clear indication that acceleration is gradient <br> ignore minus sign <br> allow <br> $(-) 4.8$ to $(-) 5.0\left(\mathrm{~m} / \mathrm{s}^{2}\right)$ | 3 |
| (ii) | clear indication that distance is area under line; understanding braking distance is area of triangle section only; evaluation; <br> e.g. <br> distance $=$ area <br> distance $=0.5 \times 30 \times 6.2$ <br> (distance $=$ ) $93(\mathrm{~m})$ | $54(\mathrm{~m})=1$ mark <br> $147(m)=2$ marks <br> accept alternative method using ecf answer from (a)(i) and $\mathrm{v}^{2}=\mathrm{u}^{2}+2$ as giving 93.75 (m) | 3 |
| (iii) | thinking distance: <br> increase in thinking distance; <br> (due to) increased reaction time; <br> braking distance: <br> no effect on braking distance; <br> (due to) no effect on braking time / braking force; | allow idea that braking distance does not depend on human factors | 4 |
| (b) | A; |  | 1 |
|  | $B$ is incorrect because it does not show deceleration $C$ is incorrect because the distance cannot change abruptly and the car is moving throughout D is incorrect because the first portion shows that the car is not moving |  |  |



Total for Question 5 = 14 marks

\begin{tabular}{|c|c|c|c|}
\hline \& Answer \& Notes \& Marks \\
\hline 6 (a) \& \begin{tabular}{l}
any four from: \\
MP1. water near heater is heated; MP2. (heated) water expands; \\
MP3. density of (heated) water decreases; MP4. lower density / warm water rises; MP5. cooler / denser water sinks; MP6. process repeats / is continuous;
\end{tabular} \& \begin{tabular}{l}
allow clear annotations on diagram \\
accept 'particles move apart from each other'/'particles spread out' reject particles expand
\end{tabular} \& 4 \\
\hline \begin{tabular}{l}
(b) (i) \\
(ii)
\end{tabular} \& \begin{tabular}{l}
temperature increases with time; idea that rate of temperature increase reduces; \\
temperature rise is quicker when container is empty; \\
with any two explanations from: \\
- particles move around quicker/have more KE in gases; \\
- convection current is faster in gases; \\
- energy transfer (by convection) is quicker; \\
- mass of air in empty container less than mass of water in full container; \\
- specific heat capacity of air is lower than water;
\end{tabular} \& \begin{tabular}{l}
allow 'temperature increases at a decreasing rate'/ EQ for 2 marks \\
allow empty container reaches higher temperature ignore comments about conduction allow particles in gases are more free to move \\
allow less particles in empty container allow RA
\end{tabular} \& 2

3 <br>
\hline
\end{tabular}

Total for Question 6 = 9 marks

| Question number | Answer | Notes | Marks |
| :---: | :---: | :---: | :---: |
| 7 (a) | any two from: <br> MP1. alphas do not penetrate as far; <br> MP2. alphas are more ionizing; <br> MP3. alphas are more likely to collide (with material); <br> MP4. alphas have more mass / move slower; | allow RA <br> allow RA <br> allow RA <br> allow RA | 2 |
| (b) $\begin{aligned} & \text { (i) } \\ & \text { (ii) } \\ & \\ & \text { (iii) }\end{aligned}$ | (nuclei with) same numbers of protons; <br> (nuclei with) different numbers of neutrons; | allow (nuclei with) same atomic number allow (nuclei with) different mass number | 2 |
|  | one mark for each correct number; |  | 2 |
|  | ${ }_{92}^{235} \cup \rightarrow \underset{90}{231} \mathrm{Th}+\stackrel{4}{2}_{{ }_{2}} \alpha$ |  |  |
|  | any indication that 2100 million years is 3 half-lives; evaluation of number of uranium nuclei after 1 half-life; | 3200 (million) | 5 |
|  | evaluation of number of uranium nuclei after 1 half-life; <br> (after 2100 million years) there are 800 million uranium nuclei; (after 2100 million years) there are 5600 million thorium nuclei; $5600 \text { (million) / } 800 \text { (million) = 7; }$ | 3200 (million) <br> uranium nuclei after one half-life scores first three marks allow total number of nuclei is constant allow $7 \times 800=5600$ |  |

Total for Question 7 = 11 marks

| Question number | Answer | Notes | Marks |
| :---: | :---: | :---: | :---: |
| 8 (a) | two magnets with opposite poles facing; poles brought close to each other; | accept annotated diagram for either mark accept from diagram if magnets drawn closer together than the thickness of their poles i.e. $A<B$ (on diagram) | 2 |
| (b) (i) <br> (ii) | any three from: <br> MP1. ammeter displays a current/reading (in one direction); <br> MP2. (because) a voltage is induced; <br> MP3. induced voltage gives a current; <br> MP4. ammeter reading goes negative / shows current in opposite direction; <br> MP5. (because) wire changes direction; <br> any four from: <br> MP1. ammeter reading / current is greater (than before); <br> MP2. because wire moving quicker; <br> MP3. more field lines cut (per second); <br> MP4. gives a larger induced voltage; <br> MP5. frequency of alternating current increases; | allow 'an alternating current / a.c. is produced' <br> allow idea that direction of current changes more frequently | $3$ $4$ |

Total for Question $8=9$ marks

\begin{tabular}{|c|c|c|c|}
\hline Question number \& Answer \& Notes \& Marks \\
\hline 9 (a) \& \begin{tabular}{l}
use of \(v^{2}=u^{2}+2 a s ;\) \\
substitution; rearrangement; evaluation; \\
e.g.
\[
\begin{aligned}
\& v^{2}=u^{2}+2 a s \\
\& v^{2}=(0)+2 \times 10 \times 2.2 \\
\& v=\int 44 \\
\& (v=) 6.6(\mathrm{~m} / \mathrm{s})
\end{aligned}
\]
\end{tabular} \& \begin{tabular}{l}
seen anywhere in working allow use of \(\mathrm{g}=9.8,9.81\) \\
allow alternative method using \(\mathrm{mgh}=1 / 2 \mathrm{mv}^{2}\) final answer of \(44(\mathrm{~m} / \mathrm{s})\) is 2 marks only
\[
6.56 \ldots(\mathrm{~m} / \mathrm{s})
\] \\
6.5 scores 3 marks only
\end{tabular} \& 4 \\
\hline \begin{tabular}{l}
(b) (i) \\
(ii)
\end{tabular} \& \begin{tabular}{l}
vertical arrow drawn upwards; \\
substitution into \(F=m a ;\) rearrangement; evaluation; \\
e.g.
\[
18000=4100 \times a
\] \\
\(a=18000 / 4100\) \\
\((\mathrm{a}=) 4.4\left(\mathrm{~m} / \mathrm{s}^{2}\right)\)
\end{tabular} \& \begin{tabular}{l}
ignore labels reject if more than one arrow drawn unless resultant force is clearly labelled \\
-1 for POT error \\
allow 4.39...(m/s \({ }^{2}\) )
\end{tabular} \& 1

3 <br>
\hline
\end{tabular}

Total for Question 9 = 8 marks

\begin{tabular}{|c|c|c|c|}
\hline Question number \& Answer \& Notes \& Marks \\
\hline 10 (a) \& ```
use of p = h }\times\mathrm{ density }\timesg\mathrm{ ;
conversion of 57 cm into 0.57 m;
evaluation;
e.g.
pressure difference = 57 \times 820 × 10
pressure difference = 0.57 \times 820 × 10
(pressure difference =) 4700 (Pa)
``` \& \begin{tabular}{l}
allow mark if formula on its own is seen in working \\
allow use of \(g=9.8,9.81\) \\
470 000, 467000 , \\
467 400, 458052 , \\
458519.4 etc. score 2 marks \\
allow 4670, 4674, 4580.52, 4585.194 etc.
\end{tabular} \& 3 \\
\hline \begin{tabular}{l}
(b) (i) \\
(ii) \\
(iii)
\end{tabular} \& ```
substitution into \(\mathrm{W}=\mathrm{m} \times \mathrm{g}\);
evaluation;
correct unit;
e.g.
W = \(24 \times 10\)
( \(\mathrm{W}=\) ) 240
newtons / N
substitution into \(p=F / A\);
evaluation;
e.g.
\(\mathrm{p}=240 / 1.2\)
( \(\mathrm{p}=\) ) \(200(\mathrm{~Pa})\)
substitution into \(p=F / A\);
rearrangement;
evaluation;
e.g.
\(200=F / 4.8\)
\(\mathrm{F}=200 \times 4.8\)
( \(\mathrm{F}=\) ) \(960(\mathrm{~N})\)
``` \& \begin{tabular}{l}
no mark for formula on its own allow use of \(g=9.8,9.81\) -1 for POT error e.g. incorrectly changing kg to g mark independently \\
allow 235.2, 235.44 \\
no mark for formula on its own allow ecf from (i) \\
no mark for formula on its own allow ecf from (ii)
\end{tabular} \& 3

2
2
3 <br>

\hline (c) \& | GPE of piston $X=$ decrease; |
| :--- |
| GPE of piston $Y=$ increase; |
| chemical energy of piston $Y=$ no change; |
| kinetic energy of piston $Y=$ no change; | \& allow marks if the meaning is clear e.g. allow + , $\uparrow$ for increase etc. \& 4 <br>

\hline
\end{tabular}



